

PASKOVA, K.A.; VASIL'YEV, S.S.; KHAMO-LEYLA, M.A.; SHAYTVA'DY, I.Ya.

Study of the radiations from  $Tc^{94}$ ,  $Tc^{96}$ , and  $Zr^{87}$ . Izv. AN  
SSSR. Ser. fiz. 29 no.12:2255-2263 D '65. (MIRA 19:1)

1. Nauchno-issledovatel'skiy institut yadernoy fiziki Moskovskogo  
gosudarstvennogo universiteta.

VASIL'YEV, S.S.; MIKHALEVA, T.N.; CHUPRUNOV, D.I.

Differential cross sections of the  $Al^{27}(p, p')Al^{27}$   
reaction for levels 7 to 13 at  $E_p = 6.56$  Mev. Vest.  
Mosk. un. Ser. 3: Fiz., astron. 19 no.4:88-89 J1-Ag '64.  
(MIRA 17:10)  
1. Nauchno-Issledovatel'skiy institut yadernoy fiziki  
Moskovskogo universiteta.

VASIL'YEV, S.S.; DZHORZH, E.T.; SHAVTVALOV, L.Ya.

Study on  $\beta^+$ -spectra of  $\text{Ne}^{19}$ ,  $\text{Ge}^{67}$ ,  $\text{Sb}^{118}$  and  $\gamma$ -radiation during bombardment of  $\text{Au}^{197}$  with  $\alpha$ -particles. Zhur. eksp. i teor. fiz. 47 no.3:1164-1167 S '64. (MIRA 17:11)

1. Institut yadernoy fiziki Moskovskogo gosudarstvennogo universiteta.

VASIL'YEV, S.S.; SELIVOKHINA, M.S.

Relation between the kinetics of the electric oxidation of nitrogen and the voltammetric characteristics of the electric discharge. Zhur. fiz. khim. 38 no.2:361-367 F '64.

(MIRA 1968)

1. Moskovskiy tekhnologicheskiy institut legkoy promyshlennosti.

ABSTRACT. V.I., starshiy prepodavatel'; YARIL'YEV, S.S., doktor khim.  
nauk, prof.

Effect of the voltage of the electric field on the gluing of  
shoe materials by means of high frequency currents. Nauch.  
trudy MTILP no.24:73-75 '62. (MIRA 16:7)

1. Kafedra fiziki Moskovskogo tekhnologicheskogo instituta  
legkoy promyshlennosti.  
(Gluing) (Induction heating)

MENTSOV, V.S., starshiy prepodavatel'; LYUBOVTSOVA, M.D., inzh.;  
VASIL'YEV, S.S., doktor khim. nauk, prof.

Effect of the voltage of the electric field on the speed and  
strength of fabric gluing by means of high frequency currents.  
Nauch. trudy MTILP no.24:65-69 :62. (MIRA 16:7)

1. Kafedra fiziki Moskovskogo tekhnologicheskogo instituta  
legkoy promyshlennosti.  
(Gluing) (Induction heating)

SHEYNIS, Ye.S., kand. tekhn. nauk, dotsent; VASIL'YEV, S.S. prof., doktor  
khim. nauk

Conductivity of moist leather at high frequencies. Nauch. trudy  
MTILP no.24:118-120 '62. (MIRA 16:7)

1. Kafedra fiziki Moskovskogo tekhnologicheskogo instituta  
legkoy promyshlennosti.  
(Leather—Electric properties)

GRIGOR'YAN, G.S.[Hryhor'ian, H.S.], dots.; KISTANOV, Ya.A., dots.;  
 FEFILOV, A.I., dots.; GENKINA, L.S.[Henkina, L.S.], dots.;  
 VASIL'YEV, S.S.[Vasil'iev, S.S.], dots.; SEREBRYAKOV, S.V.,  
 prof.; DNEPROVSKIY, S.P.[Dnieprovs'kyi, S.P.], prof.;  
 PIROGOV, P.V.[Pyrohov, P.V.], dots.; GOGOL', B.I.[Hohol', BI.],  
 dots.; SMOTRINA, N.A., dots.; KULIKOV, O.G.[Kulikov, O.H.],  
 dots.; KUZIN, M.I., dots.; DEMIDYUK, V.F.[Demydiuk, V.F.], red.;  
 SKVIRSKAYA, M.P.[Skvyrs'ka, M.P.], red.; LEVCHENKO, O.K., tekhn.  
 red.; SERGEYEV, V.F.[Serhieiev, V.F.], tekhn. red.

[Soviet trade economics] Ekonomika radians'koi torhivli; p'd-  
 ruchnyk. [By] G.S.Grigor'ian ta inshi. Kyiv, Derzhpolitvydav  
 URSR, 1962. 500 p. (MIRA 16:11)

(Russia--Commerce)



VASIL'YEV, S. S.; SHAFTVALOV, L. Ya.

Radiations from  $F^{17}$ ,  $P^{30}$ ,  $Cl^{33}$ , and  $Br^{78}$ . Izv. AN SSSR. Ser.  
fiz. 16 no.12:1495-1497 D '62. (MIRA 16:1)

(Isotopes—Spectra)

AKISHIN, A.I.; ANDREYEVA, M.G.; VASIL'YEV, S.S.; ISAYEV, L.N.;  
TSEPLYAYEV, L.I.

Action of electron bombardment and glow discharge on alloyed  
secondary electron emitters. Radiotekhnika i elektronika, 8 no.2:  
288-293 F '63. (MIRA 16:2)  
(Cathodes) (Thermionic emission)

VASIL'YEV, S. S. and GRIGOR'YAN, S.S.

Ekonomika sovetskoy trgovli: uchebnik [by] G. S.  
Grigor'yan, S.S. Vasil'yev [i Dr] Red. koll. S.S.  
Vasil'yev [i Dr] Moskva, Gospolitizdat, 1962.  
527 p. tables.

10869

S/048/62/026/003/004/011  
B125/B186

24.660

AUTHORS: Vasil'yev, S. S., Romanovskiy, Ye. A., and Timushev, G. F.

TITLE: Inelastic scattering of 6.6-Mev protons from nickel and copper nuclei

PERIODIC L: Akademiya nauk SSSR.. Izvestiya. Seriya fizicheskaya, v. 26, no. 2, 1962, 1143-1149.

TEXT: The inelastic scattering of 6.6-Mev protons from  $Ni^{58}$ ,  $Ni^{60}$ ,  $Cu^{63}$  and  $Cu^{65}$  nuclei is studied in detail. The proton beam from the 120-cm cyclotron of the IILAF MU was focused into the reaction chamber by quadrupole lenses. The protons from nickel and copper foils with natural isotopic composition scattered through an angle  $\theta$ , were analyzed with a double-focusing magnetic spectrometer. The energies of the excited states as measured (Table 1) are in good agreement with the results of G. H. Paris, W. W. Buechner, Bull. Amer. Soc., Ser. II, 2, 61 (1957) and of M. Mazari et al., Phys. Rev., 108, 373 (1957). The inelastic proton scattering occurs probably via compound nucleus formation because the angular distributions of the scattered protons are isotropic within the limits of Card 1/4

Inelastic scattering of 6.6-Mev ...  
measurement error.

S/048/62/026/009/004/011  
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$$\sigma = \frac{1}{2} \pi \lambda^2 \left\{ T_0 \left[ \frac{2(2T'_2)}{T_0 + 2T'_2} \right] + T_1 \left[ \frac{2(T'_1 + T'_2)}{T_1 + T'_1 + T'_2} + \frac{4(2T'_1 + 2T'_2)}{T_1 + 2T'_1 + 2T'_2} \right] + \right. \\ \left. + T_2 \left[ \frac{10(T'_0 + 2T'_2)}{T_2 + T'_0 + 2T'_2} \right] + T_3 \left[ \frac{6(2T'_1 + 2T'_2)}{T_3 + 2T'_1 + 2T'_2} + \frac{8(T'_1 + 2T'_2)}{T_3 + T'_1 + 2T'_2} \right] + \right. \\ \left. + T_4 \left[ \frac{8(2T'_2)}{T_4 + 2T'_2} + \frac{10T'_2}{T_4 + T'_2} \right] \right\} \quad (6)$$

is the total inelastic scattering cross section of protons ( $E_p = 6.6 \text{ Mev}$ ) from  $\text{Ni}^{60}$ . The "penetrabilities"  $T_L$  and  $T_{L'}$  are equal to zero if  $T > 4$ , and  $T' > 3$ .  $L$  and  $L'$  are the orbital angular momenta of the incident and of the outgoing proton. The contribution of the *direct* processes to the scattering here considered is negligibly small. Table 2 gives the total  
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Inelastic scattering of 6.6-Mev ...

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cross sections of the proton scattering from  $\text{Cu}^{63}$  and  $\text{Cu}^{65}$  with  $E_p = 6.6 \text{ Mev}$ . The sum of the total cross sections of inelastic scattering in the range up to 3.5 Mev is  $230 \pm 20$  millibarn for  $\text{Ni}^{58}$  and  $250 \pm 30$  millibarn for  $\text{Ni}^{60}$ . For this reason the fraction of the pp-processes that occurs via a compound nucleus formation may be 300 to 350 millibarn in the scattering of protons from  $\text{Ni}^{58}$  and  $\text{Ni}^{60}$  at  $E_p = 6.6 \text{ Mev}$ . The present results do not contradict the hypothesis of increased elastic scattering cross section of even-even  $\text{Ni}^{58}$  and  $\text{Ni}^{60}$  nuclei through large angles due to the great contribution of the pp-processes taking place via a compound nucleus. There are 6 figures and 2 tables.

ASSOCIATION: Nauchno-issledovatel'skiy institut yadernoy fiziki  
Moskovskogo gos. universiteta im. M. V. Lomonosova  
(Scientific Research Institute of Nuclear Physics of the  
Moscow State University imeni M. V. Lomonosov)

Card 3/4

Inelastic scattering of 6.6-Mev ...

S/048/62/026/009/004/011  
B125/B186

Table 1. Energies of the excited states in  $\text{Ni}^{58}$ ,  $\text{Ni}^{60}$ ,  $\text{Cu}^{63}$ , and  $\text{Cu}^{65}$  (in Mev).

Legend to Table 2: (1) process; (2)  $\sigma$  total

$\text{Ni}^{58}$	$\text{Cu}^{63}$	$\text{Cu}^{65}$
$1,450 \pm 0,006$	$0,657 \pm 0,007$	$0,777 \pm 0,008$
$2,457 \pm 0,012$	$0,856 \pm 0,008$	$1,108 \pm 0,008$
$2,772 \pm 0,012$	$1,328 \pm 0,015$	$1,480 \pm 0,010$
$2,892 \pm 0,012$	$1,419 \pm 0,015$	$1,635 \pm 0,015$
$2,941 \pm 0,012$	$1,544 \pm 0,015$	$1,730 \pm 0,015$
$3,036 \pm 0,012$	$1,858 \pm 0,015$	$2,000 \pm 0,015$

(1) Процесс	(2) $\sigma$ , Mb	
	$\text{Cu}^{63}$	$\text{Cu}^{65}$
(p, p')	$240 \pm 30$	$70 \pm 15$
(p, $\alpha$ )	$35 \pm 3$	$37 \pm 10$
(p, n)	$300 \pm 30$	$500 \pm 50$
(p, p)	$\sim 25$	$\sim 10$
(2) $\sigma_{\text{total}}$	$600 \pm 50$	$617 \pm 50$

Card 4/4

S/056/62/042/002/015/055  
B102/B136

AUTHORS: Vasil'yev, S. S., Romanovskiy, Ye. A., Timushev, G. F.  
TITLE: Inelastic scattering of 6.6-Mev protons from  $\text{Ca}^{40}$  and  $\text{Mn}^{55}$  nuclei  
PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v 42, no. 2 1962, 395-402

TEXT: Inelastic proton scattering was studied with a rotating magnetic analyzer. The protons were accelerated in the 120-cm cyclotron of the Institute of Nuclear Physics of MGU [Assoc.]. The angular distributions were measured, of elastically scattered protons or proton groups corresponding to the excited levels  $3.352 \pm 0.010$ ,  $3.733 \pm 0.014$  and  $3.912 \pm 0.015$  Mev of  $\text{Ca}^{40}$  and  $0.131 \pm 0.007$ ,  $0.984 \pm 0.005$ ,  $1.291 \pm 0.010$ ,  $1.523 \pm 0.007$  and  $1.885 \pm 0.007$  Mev of  $\text{Mn}^{55}$ . The scattering mechanism is most probably a (p,n) reaction; its threshold is at 15 Mev for  $\text{Ca}^{40}$  and 1.020 Mev for  $\text{Mn}^{55}$ . From the results shown in Card 1/3



Inelastic scattering of...

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diagrams, it can be seen that all proton angular distributions are anisotropic and asymmetric with respect to the  $90^\circ$  angle, except for proton scattering from  $Mn^{55}$  when exciting the 1.523-Mev level. This distribution is isotropic according to the statistical theory. These results indicate that direct excitation is the main mechanism in inelastic proton scattering from  $Ca^{40}$  and  $Mn^{55}$ . This conclusion was verified by comparing the results with theoretical ones obtained with the model of direct interactions in inelastic scattering. Spin and parity of the levels considered were determined for some cases. For the 3.912-Mev level of  $Ca^{40}$ ,  $2^+$  was obtained, and  $3^-$  for the 5.733-Mev level, which gives the sequence  $0^+$ ,  $3^-$ ,  $2^-$  for the lowest  $Ca^{40}$  levels and agrees with the results of other authors. For the  $Mn^{55}$  levels 0.131, 0.984, 1.291 and 1.523 Mev,  $(7/2)^+$ ,  $(9/2)^+$ ,  $(11/2)^+$ ,  $(3/2)^+$  are obtained, respectively, and  $(7/2)^+$  is most probable for the 1.885-Mev level. These characteristics show that collective and single-particle excited states exist in the  $Mn^{55}$  nucleus. Yu. A. Verobiyev, A. A. Danilov, Ye. F. Kir'yanov, V. P. Khlapov, Z. F. Kalacheva, M. Kh. Listov, R. I. Osipova, T. I. Dyukova and L. P. Kovaleva are thanked for help. A. K. Valster, I. I. Zalyubovskiy.

Card 2/5

Inelastic scattering of...

S/056/62/042/002/015/055  
B102/B:38

V. P. Lutsik (UFZh. 4, 705, '959) and A. V. Luk'yanov, I. B. Teplov, M. K. Akimova (Tablitsa volnovykh kulonovskikh funktsiy - Tables of Coulomb wave functions - Izd. AN SSSR, 1961) are mentioned. There are 6 figures and 23 references: 8 Soviet and 15 non-Soviet. The four most recent references to English-language publications read as follows: A. M. Lane, E. D. Pendlebury, Nucl. Phys. 5, 39, '960; G. E. Browne et al. Nucl. Phys. 24, 1, 1961; N. Nath et al. Nucl. Phys. 13, 74, '959; E. Post, N. Austern, Phys. Rev. 120, 1375, 1960.

ASSOCIATION: Institut yadernoy fiziki Moskovskogo gosudarstvennogo universiteta (Institute of Nuclear Physics of Moscow State University) ✓

SUBMITTED: August 31, '96

Card 3/3

S/056/62/042/002/018/055  
B102/B138

AUTHORS: Baskova, K. A., Vasil'yev, S. S., No Seng Ch'ang, Shavtvalov,  
L. Ya.

TITLE: Investigation of some radioactive nuclei in the range of  
filled  $1f_{7/2}$  shells

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 42,  
no. 2, 1962, 416-426

TEXT: A magnetic thin-lens  $\beta$ -spectrometer and a scintillation  $\gamma$ -spec-  
trometer were used to investigate the radiation emitted by  $Ni^{65}$ ,  $Co^{55}$ ,  
 $Mn^{51}$ ,  $V^{47}$ , and  $Se^{83}$  nuclei. These isotopes were produced by proton or  
deuteron irradiation of enriched targets in the cyclotron of the NIIYaF MGU.  
The following results were obtained: 2.5-hr  $Ni^{65}$  was produced in the reac-  
tion  $Ni^{64}(d,p)Ni^{65}$ . In the  $Ni^{65}$  spectrum three partial  $\beta^-$ -transitions  
with  $2120 \pm 40$ , 1050 and 620 keV end-point energies (intensities 57, 14 and  
29%) and 370, 1120, 1490, 1630 and 1720 keV  $\gamma$ -transitions were observed.  
 $\beta^- \gamma$ -coincidences were observed at 1490 keV and 1120 keV gamma energies.

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Investigation of some radioactive ...

18-hr  $\text{Co}^{55}$  was produced in the reaction  $\text{Fe}^{54}(\text{d},\text{n})\text{Co}^{55}$ ; the end-point energies of the three  $\beta^+$ -spectrum components were  $1500 \pm 30$ , 1040 and 550 kev (56, 41, 3%), gamma lines were observed at 940, 1410, 1800 and 2180 kev. The  $\beta^+$ -transition with the end-point energy 1500 kev takes place to an excited level with subsequent emission of 940-kev gamma rays.  $\beta^-$  coincidence was observed for 1410 and 940 gamma quanta, the end-point energy of the  $\beta^+$  particles was 1040 kev. The 44-min  $\text{Mn}^{51}$  was obtained from  $\text{Cr}^{50}(\text{d},\text{n})\text{Mn}^{51}$  reactions. The end-point energy of the two  $\beta^+$  spectrum components are at 600 and at  $2170 \pm 60$  kev, in the  $\gamma$ -spectrum hitherto unknown lines were observed at 1560 and 2030 kev, with a half-life of  $50 \pm 10$  min. The 1560-kev transition is assumed to follow the 600-kev  $\beta^+$ -decay, the 1569 and 2030-kev levels belong to the reaction  $\text{V}^{51}(\text{p},\text{n})\text{Cr}^{51}$ . The 33-min  $\text{V}^{47}$  isotope was obtained from  $\text{Ti}^{47}(\text{p},\text{n})\text{V}^{47}$ . It is shown a simple  $\beta^+$  spectrum with an end-point energy of  $1890 \pm 30$  kev, gamma lines were observed at 1800 and 2160 kev, the latter unknown up to now. The 25-min  $\text{Se}^{83}$  was produced by a (d,p) reaction from  $\text{Se}^{82}$ . Three  $\beta^-$  components were found with 1.0, 1.8 and 3.3 Mev end-point energies

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Investigation of some radioactive ...

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(58, 40, ~2%); the latter is a new. Gamma transitions were recorded at 220, 355, 530, 780, 1060, 1300, 1480, 1850 and 2300 kev. Only those with 220, 355, 1850 and 2300 kev belonged to the 25-min activity, the others to  $\text{Br}^{82}$ . The results are discussed on the assumption that one group of the odd nuclei investigated had one nucleon outside the filled  $1f_{7/2}$  shell, and in the other group one nucleon is deficient to fill this shell. Nuclei with 29 p or n have similar excited levels at ~600, 1000 and 1400 kev, those with 27 p or n only at ~1400 kev. The excitation energy decreases with increasing number of even p and increases with the number of even n. The configurations of the ~1400-kev levels will be  $(1f_{7/2})^{-1}(2p_{3/2})^2$  for  $Z(N) = 29$  and  $(1f_{7/2})^{-2}(2p_{3/2})^1$  for  $Z(N) = 27$ . ✓

Yu. A. Vorob'yev, V. S. Zazulin, A. A. Vasil'yev, and I. Ya. Ushakov are thanked for help. There are 16 figures, 1 table, and 22 references: 2 Soviet and 20 non-Soviet. The four most recent references to English-language publications read as follows: L. H. Th. Rietjens et al. Phys. Rev. 120, 527, 1960; M. K. Ramaswamy et al. Proc. Intern. Conf. Nucl. Struc. Canada, 1960, p. 963. R. W. Bauer, M. Deutsch. Nucl. Phys. 16, 264, Card 3/0 4

Investigation of some radioactive ...

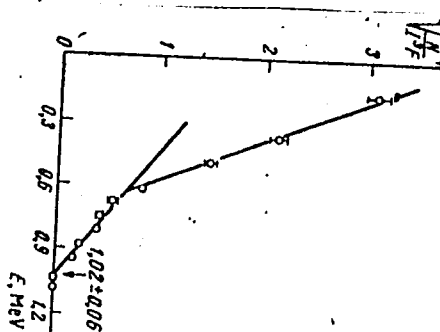
S/056/62/042/002/018/055  
B102/B138

1960; M. Nozawa et al. J. Phys. Soc. Japan, 15, 2137, 1960.

ASSOCIATION: Institut yadernoy fiziki Moskovskogo gosudarstvennogo universiteta (Institute of Nuclear Physics of Moscow State University)

SUBMITTED: September 23, 1961

Fig. 13



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VASIL'YEV, S.S.; KOMAROV, V.V.; POPOVA, A.M.

Properties of the lower states of  $\text{Li}^5$  and  $\text{Be}^8$  nuclei  
appearing in the decay of light nuclei. Izv. AN SSSR.  
Ser. fiz. 25 no.9:1117-1120 '61. (MIRA 14:8)

1. Institut yadernoy fiziki Moskovskogo gosudarstvennogo  
universiteta imeni M.V. Lomonosova.

(Lithium)

(Beryllium)

(Nuclear reactions)

VASIL'YEV, S.S.; NON SEN CHAN; SHAVTVALOV, L.Ya.

Investigating the radiation from  $Zn^{63}$ . Zhur. eksp. i teor.  
fiz. 40 no.2:475-476 F '61. (MIRA 14:7)

1. Institut yadernoy fiziki Moskovskogo gosudarstvennogo  
universiteta.  
(Zinc—Istopes) (Radiation)



18.8/00

31518

S/058/61/000/010/089/100  
A001/A101

AUTHORS: Artseishevskiy, M.A., Vasil'yev, S.S., Koshelyayev, G.V., Selisskiy, Ya.P.

TITLE: Effect of deuteron irradiation on electric resistance of ordering and aging alloys

PERIODICAL: Referativnyy zhurnal. Fizika, no. 10, 1961, 279, abstract 10E434  
("Sh. tr. Tsentr. n. 1. in-t chernoy metallurgii", 1959, no. 22, 168 - 176)

TEXT: The authors investigated the effect of irradiation by 4-Mev deuterons on electric resistance of the ordering alloys  $\text{Ni}_3\text{Fe}$ ,  $\text{Fe}_3\text{Al}$  and the aging alloy 35% Ni, 4.5% Ti, the rest being Fe. In all cases irradiation by beams of up to  $5 \times 10^{17}$  deuteron/cm<sup>2</sup> caused sharp changes of electric resistance: in the ordered alloy  $\text{Fe}_3\text{Al}$  it grew considerably, in the annealed one it dropped, in  $\text{Ni}_3\text{Fe}$  in the ordered and disordered states electric resistance decreased sharply. In the Fe-Ni-Ti alloy the effect was not greater than experimental errors. The most probable process causing decrease of electric resistance is ordering. As a result

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S/058/61/000/010/089/100  
A001/A101

Effect of deuteron irradiation...

of irradiation of the ordered  $\text{Fe}_3\text{Al}$ , some intermediate degree of ordering is possibly attained. This hypothesis is confirmed by the character of changes in electric resistance of the alloy at annealing which was carried out with specimens after measurements in the irradiated state.

V. Patskevich

[Abstracter's note: Complete translation]

X

Card 2/2

24-6500

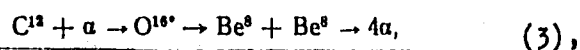
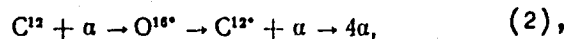
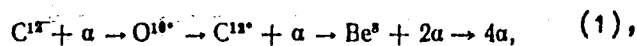
31771  
S/056/61/041/006/011/054  
B113/B104

AUTHORS: Vasil'yev, S. S., Komarov, V. V., Popova, A. M.

TITLE: Investigation of the reaction  $C^{12}(\alpha, 4\alpha)$

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 41, no. 6(12), 1961, 1757-1760

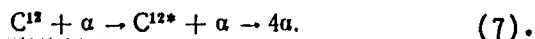
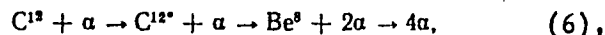
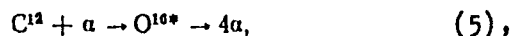
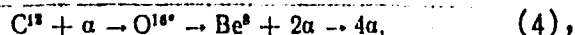
TEXT: The authors studied the decay of the  $C^{12}$  nucleus into 3  $\alpha$ -particles induced by a 23-Mev  $\alpha$ -particle. The  $\alpha$ -particles were accelerated on the 120-cm cyclotron of the NIIYaF NGU, the reactions took place in НИКФИ (NIKFI)-type nuclear emulsion plates of 50-400  $\mu$  thickness: Я 2 (Ya2), T-1 (T-1), T-2 (T-2), T-3 (T-3), and D (D). The following mechanisms are possible:



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Investigation of the reaction ...

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To determine the probability of these reaction modes, the authors studied the excitation energy of the  $C^{12}$  and  $Be^8$  compound nuclei, the angular and energy distribution of the  $\alpha$ -particles. The weight of the true values  $E_{exc}(C^{12})$  must amount to  $1/4$  if the reaction proceeds according to mechanisms (1) or (2). The distribution of the calculated values  $E_{exc}(C^{12})$  was also measured. Mechanisms (1) and (2) proved to be very unlikely. The probability of the modes (3), (4), (5) in the decay of the  $C^{12}$  nucleus is determined from the energy distribution of the resulting  $\alpha$ -particles. If the reaction proceeds through a straight decay of the  $O^{16*}$  compound nucleus into four independent particles, the energy distribution of the resulting  $\alpha$ -particles must satisfy the formula:

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S/056/61/041/006/011/054  
B113/B104

$F(E_\alpha) = E_\alpha^{1/2} \cdot (E_{\max} - \mu \cdot E_\alpha)^{1/2}$ . A comparison of the curve obtained from this formula with the curve for the case where the compound nucleus  $O^{16*}$  decays into two  $Be^8$  nuclei, and the latter into two  $\alpha$ -particles each, showed that mode (3) was unlikely. Most probable is the formation of an  $O^{16*}$  compound nucleus decaying into four  $\alpha$ -particles with resonance interaction between the  $\alpha$ -particles in the ground state. There are 2 figures and 5 Soviet references.

SUBMITTED: June 27, 1961

X

Card 3/3

MENTSOV, V.S., inzh.; VASIL'YEV, S.S., doktor khimicheskikh nauk, prof.

Studying the kinetics of gluing samples of unbleached cotton fabrics  
with casein adhesives. Izv. vys. ucheb. zav.; tekhn. leg. prom.  
no. 1:71-79 '60. (MIRA 14:5)

1. Moskovskiy tekhnologicheskii institut legkoy promyshlennosti.  
Rekomendovana kafedroy fiziki.  
(Adhesives) (Cotton fabrics)



Study of  $Zn^{63}$  emission ...

S/056/61/040/002/013/047  
B102/B202

relative intensities of these spectra are the following: 2, 10, 10, 10, and 68. The gamma spectrum was studied by means of a luminescence spectrometer. The pulses from the photomultiplier  $\Phi\Phi\Phi-1C$  (FEU-1S) (with NaI(Tl) crystal) were fed into a 100-channel pulse height analyzer of the type AM-100(AI-100). The gamma spectrum is shown in Fig. 2 (abscissa: number of channels). Besides the intense annihilation peak, the following lines were recorded:  $680 \pm 10$ ,  $910 \pm 10$ ,  $1350 \pm 20$ ,  $1430 \pm 20$ , and  $2300 \pm 30$  kev. The existence of a line with  $1540 \pm 20$  kev is also probable. The half-life was found to be  $37.6 \pm 0.3$  min. The results are in good agreement with those obtained by Ricci et al. (Nuovo Cim. 11, 156, 1959) but not with those obtained by Huber et al. (Helv. Phys. Acta, 20, 495, 1947). There are 2 figures and 4 references: 1 Soviet-bloc and 3 non-Soviet-bloc.

ASSOCIATION: Institut yadernoy fiziki Moskovskogo gosudarstvennogo universiteta (Institute of Nuclear Physics of Moscow State University)

Card 2/3



Study of  $Zn^{63}$  emission ...

SUBMITTED: September 12, 1960

S/056/61/040/002/013/047  
3102/3202

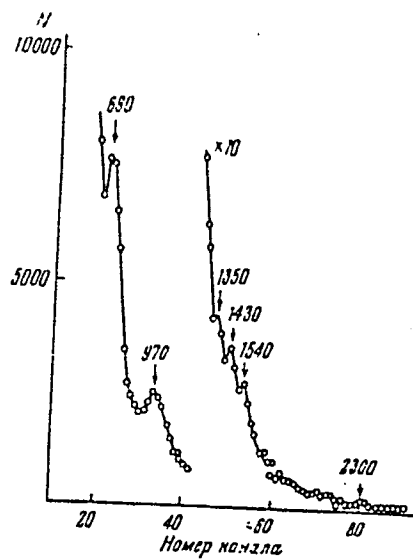


Рис. 2

Card 3/3

VASIL'YEV, S.S.; KOMAROV, V.V.; POPOVA, A.M.

Energy state of the  $\text{Be}^8$  nucleus in the decay reaction of the  $\text{C}^{12}$  nucleus into three  $\alpha$ -particles under the effect of protons and neutrons. Izv.AN SSSR.Ser.fiz. 24 no.9:1149-1152 S '60.

(MIRA 13:9)

1. Nauchno-issledovatel'skiy institut yadernoy fiziki Moskovskogo gosudarstvennogo universiteta im. M.V.Lomonosova.

(Beryllium--Isotopes)

(Carbon--Decay)

S/056/60/039/005/007/051  
B029/B077

AUTHORS: Vasil'yev, S. S., Shaftvalov, L. Ya.

TITLE: The  $\beta^+$  Spectrum of  $\text{Si}^{27}$

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1960,  
Vol. 39, No. 5(11), pp. 1221 - 1223

TEXT: The authors investigated the  $\beta^+$  spectrum of  $\text{Si}^{27}$  using a  $\beta$ -spectrometer with a magnetic lens.  $\text{Si}^{27}$  was obtained by the reaction  $\text{Al}^{27}(\text{p}, \text{n})\text{Si}^{27}$  in the 120-cm cyclotron of NIIYaF MGU (Scientific Research Institute of Nuclear Physics, Moscow State University). The proton beam emitted by the cyclotron was focused by quadrupole lenses and hit the target which was placed 9 m away from the cyclotron behind a concrete shield. Before hitting the target the protons passed through a screen connected to an integrator. The aluminum target consisted of a 2.7 mg/cm<sup>2</sup> thick rotating ring. This arrangement brought the irradiated parts of the target into the focus of the  $\beta$ -spectrometer, and using a

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The  $\beta^+$  Spectrum of  $\text{Si}^{27}$ S/056/60/039/005/007/051  
B029/B077

suitable rotation velocity it was possible to move the radioisotopes into the focus of the  $\beta$ -spectrometer with the wanted half-life. A strong background complicated the measurements considerably. The Fermi chart shows that the  $\beta^+$  spectrum of  $\text{Si}^{27}$  consists of two partial spectra. The upper limit of the fundamental  $\beta^+$  spectrum is  $3.65 \pm 0.05$  Mev, and its relative intensity is  $\sim 90\%$ . The upper limit of the weaker  $\beta^+$  spectrum is  $1.45 \pm 0.1$  Mev, and its relative intensity is  $< 10\%$ . The decay scheme found in the book of B. S. Dzhelepov and L. K. Peker (Ref.5) yields a level of 2270 kev in  $\text{Al}$  if a partial  $\beta^+$  spectrum with an upper limit of  $1.45 \pm 0.1$  Mev is added. This level is observed when investigating inelastic scattering. It has a positive parity and a  $5/2$  spin like the ground state of the  $\text{Si}^{27}$  nucleus. Therefore, the  $\beta$ -transition that leads to this level is more probable than transitions leading to other states of  $\text{Al}^{27}$ . The half-life of  $\text{Si}^{27}$  was calculated to be  $4.1 \pm 0.4$  sec and agrees with the known value within the limits of error. Finally, the method of half-life determination is briefly described. The authors thank the cyclotron team and

Card 2/3

The  $\beta^+$  Spectrum of Si<sup>27</sup>

S/056/60/039/005/007/051  
B029/B077

especially Yu. A. Vorob'yev, Z. I. Tikhomirova, B. M. Makuni, and N. S. Kirpichev for their cooperation, and also B. S. Zazulin for calculating the half-life. There are 3 figures and 6 references: 2 Soviet, 3 US, and 1 Canadian.

ASSOCIATION: Institut yadernoy fiziki Moskovskogo gosudarstvennogo universiteta (Institute of Nuclear Physics, Moscow State University)

SUBMITTED: June 23, 1960

Card 3/3

BORISOV, Gleb Borisovich; VASIL'YEV, Sergey Vasil'yevich

[Sverdlov Machine-Tool Plant; an outline history of the  
Sverdlov Machine-Tool Plant in Leningrad, 1867-1961] Stan-  
kostroitel'nyy imeni Sverdlova; ocherk ; ocherk istorii Le-  
ningradskogo stankostroitel'nogo zavoda imeni Sverdlova, 1867-  
1961 gg. Leningrad, Lenizdat, 1962. 350 p. (MIRA 16:1)  
(Leningrad--Factories)

VASIL'YEV, S. V.

PA 15/49T75

USSR/Engineering  
Peat Industry  
Peat - Production

Jun 48

"Problems of Glavtorfostroy in 1948," S. V. Vasil'yev,  
Dir Glavtorfostroy, 2 3/4 pp

"Torf Prom" No 6

Summarizes Glavtorfostroy program for 1948.

15/49T75

VASIL'YEV, S. V.

422P8: VASIL'YEV, S. V. - Stroit' luchshe, vsytsee i deshevle. (Glasnost' i pravda). Tserf. prom-st'. 1948, No 11, s 11-14.

SC: Letopis' Zhurnal'nykh Stroy, Vol. 47, 1948.



VASIL'YEV, S. V.

28999 Za povyshenie Kachestvennykh pokazateley Raboty i orfyanykh stroitel'stv.  
Torf. prom-st', 1949, No. 9, S. 1-4

SO: Letopis' Zhurnal'nykh Statey, Vol. 39, Moskva, 1949

1. VASIL'EV, S. V.
2. USSR (600)
4. Peat Industry
7. Lowering construction costs is the most important work in building peat enterprises. Torf. prom. 29 no. 10. '52.
9. Monthly List of Russian Accessions, Library of Congress, February 1953. Unclassified.

VASIL'YEV, Sergey Vladimirovich; ROZENBERG, Boris Ivanovich; KONDAKHCHAN,  
V.S., redaktor; VOHONIN, K.P., tekhnicheskii redaktor.

[Electrification of peat-extracting enterprises] Elektrifikatsiia  
torfopredpriatii. Moskva, Gos. energ. izd-vo, 1954. 360 p.  
(Peat machinery) (MLRA 8:2)

VASIL'YEV, S.V., inzh., red.; KOPEYKINA, L.V., red.; FRIDKIN,  
L.M., tekhn. red.

[Present state and measures for the further improvement of industrial safety and safety engineering on construction sites, enterprises, and in organizations of the State Production Committee on Power Engineering and Electrification of the U.S.S.R.; collection of papers presented at a conference in Moscow on July 27-30 1962] O sostoianii i merakh po dal'-neishemu uluchsheniiu okhrany truda i tekhniki bezopasnosti na stroikakh, predpriyatiyakh i v organizatsiyakh Gosudarstvennogo proizvodstvennogo komiteta po energetike i elektrifikatsii SSSR; sbornik materialov soveshchaniia, 27-30 iuliia 1962 g. Moskva, Gosenergoizdat, 1963. 190 p.

(MIRA 17:3)

1. Soveshchaniye po okhrane truda i tekhnike bezopasnosti na stroykakh i predpriyatiyakh ministerstva stroitel'stva elektrostantsii SSSR, Moscow, 1962.

SOV/81-59-16-57064

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 16, pp 165-166 (USSR)

AUTHOR: Vasil'yev, G.V.

TITLE: The Action of Hyponitric Anhydride on Aliphatic Unsaturated Ketones

PERIODICAL: Tr. Mosk. in-ta tonkoy khim. tekhnol., 1958, Nr 8, pp 47-55

ABSTRACT: The action of  $N_2O_4$  on unsaturated ketones has been studied with the aim of obtaining nitroketones and aminoketones. It has been found that in the presence of a double bond in the C-atoms which are adjacent to the CO-group, the ketones produce addition products which are unstable in an acidic medium. In proportion to the distance of the double bond from the CO-group stabler addition products are obtained. The  $NO_2$ -group is added to the least and the ONO-group to the most hydrogenated C atom. The  $NO_2$ -group in nitroketones is split off more difficultly than in nitroacids. 18 g  $CH_3COCH=CH_2$  in 150 ml petroleum ether are saturated at  $-2^\circ C$  within 5 hours by 24 g  $N_2O_4$ , and 30 g  $CH_3COCH(NO_2)CH_2ONO$  (I) are separated which is decomposed during distillation, at heating with water and in storing. On shaking I with water, alkalis and mineral acids, the ONO-group is substituted by OH with the formation of  $CH_3COCH(NO_2)CH_2OH$  (II),

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SOV/81-59-16-57064

The Action of Hyponitric Anhydride on Aliphatic Unsaturated Ketones

$n_D^{22}$  1.4512,  $d_4^{20}$  1.2182. In the reduction of Sn in concentrated HCl, II yields  $\text{CH}_3\text{COCH}_3$ ,  $\text{NH}_4\text{OH}$  and  $\text{NH}_4\text{Cl}$ . 50 g  $(\text{CH}_3)_2\text{C}=\text{CHCOCH}_3$  in 250 ml ether is saturated by 60 g  $\text{N}_2\text{O}_4$  at  $-1^\circ\text{C}$ , and 94.9 g  $(\text{CH}_3)_2\text{CNO}_2\text{CH(ONO)COCH}_3$  (III) is separated. Heating of 10 g of II in 100 ml water for 4 hours yields 3.8 g  $(\text{CH}_3)_2\text{CNO}_2\text{CHOHCOCH}_3$ , b. p. 119 -  $120^\circ\text{C}/20$  mm,  $n_D^{20}$  1.4578,  $d_4^{20}$  1.187; benzoyl derivative, melting point  $270^\circ\text{C}$  (decomposes; from alcohol). To a mixture of 50 g of III, 50 ml alcohol and 200 g Sn, HCl acid is added at  $\sim 20^\circ\text{C}$ , the neutral products are distilled by water steam, diluted by water, treated by  $\text{H}_2\text{S}$ , the filtrate is concentrated, and 6.7 g  $\text{CH}_3\text{COCHOHC(CH}_3)_2\text{NH}_2 \cdot \text{HCl}$  is extracted by anhydrous alcohol; oxime, m. p. 195 -  $197.5^\circ\text{C}$  (decomposes). In the saturation of 13.8 g 3-methyloctene-3-on-7 in 150 ml petroleum ether by 12 g  $\text{N}_2\text{O}_4$  (for 5 hours,  $-1^\circ\text{C}$ ), 21.8 g  $\text{CH}_3\text{CO(CH}_2)_2\text{CH(ONO)C(CH}_3)_2(\text{C}_2\text{H}_5)\text{NO}_2$  (IV) is formed,  $n_D^{20}$  1.4550,  $d_4^{20}$  1.2032. The treatment of IV by water yields  $\text{CH}_3\text{CO(CH}_2)_2\text{CH(OH)C(CH}_3)_2(\text{C}_2\text{H}_5)\text{NO}_2$  (V), m. p. 164 -  $165^\circ\text{C}$  (from alcohol ethylacetate); acetyl derivative, m. p. 173 -  $175^\circ\text{C}$ . The reduction of 10 g of V, by 50 g Sn and 150 ml HCl (each 15 - 20 ml after 10 - 15 min) after 72 hours leads to 1.6 g  $\text{CH}_3\text{CO(CH}_2)_2\text{CHOHC(CH}_3)_2(\text{C}_2\text{H}_5)\text{NH}_2 \cdot \text{HCl}$ , m. p. 261 -  $264^\circ\text{C}$  (decomposes; from alcohol); oxime, m. p. 298 -  $301^\circ\text{C}$  (decomposes).

V.T.

Card 2/2

AFANAS'YEV, S.A.; VASIL'YEV, S.V. (g. Cheboksary)

Electrolytic production of the metallic calcium from fused nitrates.  
Khim. v shkole 13 no.4:29-30 JI-Ag '58. (MIRA 11:6)  
(Calcium)

VASII'YEV, S. V., Engineer

\*"Rolling-Out of Minethrower Tubes", Stanki I Instrument, 14, No. 4-5, 1943.

BR-52059019

\*EXcerpt from his report:



1ST AND 2ND DEGREE																										3RD AND 4TH DEGREE																									
PROCESS AND PROPERTIES INDEX																																																			
<p><i>Ca</i> <span style="float: right;">17</span></p> <p>Analysis of the essential oil of camphor basil (<i>Ocimum</i>  <i>canum</i>) after removing camphor. 1. Kh. Fel'dman and  S. V. Vash'ev. <i>Farmatsiya</i> 4, No. 6/7, 31-3(1941).  Decamphorized oil of camphor basil still contained 6.84%  camphor along with some pinene, camphene, dipentene  and cadimene. Other components included an unsatd. alc.  <math>C_{15}H_{26}OH</math> (5.34%) and an ester <math>C_{15}H_{26}COOMe</math> (6.67%).  Julian F. Smith</p>																																																			
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**Sulfanilamide derivatives of aromatic aromatic acids**  
**I. Sulfanilamide compounds of *p*-aminobenzenesulfonic acid.** S. V. Vasil'ev (All-Union Chem. Pharm. Research Inst., Moscow). *J. Gen. Chem. (U.S.S.R.)* 10, 151 (1940). Na arsanilate (60 g.) in 300 cc. water was treated with 60 g. *p*-Cl-NHCl $\cdot$ HSO $_3$ Cl and stirred for 1.5 hrs. at 40-50° to yield, on cooling and partial concn., 31.6 g. of *p*-*N*-*arylsulfanilamidoarsenonic acid*, m.p. 240-241° (decomp. 280°) (from EtOH). The Ac. compl. was hydrolyzed by stirring with HCl (d. 1.095) at 82-85° to yield *p*-sulfanilamidoarsenonic acid (I), the const. of which are not given (there is an apparent printer's error in which at least one paragraph is omitted). 1.10 g. in 300 cc. H $_2$ O and 8 cc. 3 N NaOH was treated with 12.5 g. Na $_2$ SO $_3$ , added over a period of 1.5 hrs., after which the mixt. was stirred at 50-60° for 1 hr. to yield 7.2 g. 1,4-disulfanilamidoarsenbenzene, m. 158-60° (from EtOH). However, when the reduction was performed by SO $_2$  in NaOH soln. there was obtained *N*-(*p*-arsenophenyl)sulfanilamide, m. 205° (from dld. EtOH). The acid was inactive against spirochetes or trypanosomes, while the last 2 compds. were 24 times less active than novarsenol. G. M. Kosolapov

G. M. Kozlovskii

ASAC, SIA, POLICE, CRIMINAL, LITERATURE, CLASSIFICATION

VASIL'YEV, S. V.

Vasil'yev, S. V. - "The electrochemical reduction of amino-3-oxy-Aphenyl arsenic acid", Trudy Mosh. in-ta tekhn. tekhnologii im. Lomonosova, Issue 2, 1948, p. 23-43, - Bibliog: 6 items.

SO: U-3042, 11 March 53, (Letopis 'Zhurnal 'nykh Statey, No. 6, 1949).



( $\text{CH}_3\text{CO}_2\text{H}$ ); the residual oil after removal of the above solids had zero iodine no. and was insol. in  $\text{H}_2\text{O}$ , sol. in org. solvents; it reacted with silicotungstic acid but not with Pt chloride; its compn.,  $\text{C}_{11}\text{H}_{10}\text{NO}_4$ , identified it as an *aminomaleic acid*. The mother liquor after removal of I was chilled to  $-10^\circ$ , giving 8.3 g. solid, m.  $91-4^\circ$  (from EtOH), which explodes on strong heating but could be stored without decomn. at  $17-20^\circ$ ; its compn. was  $\text{C}_{11}\text{H}_{10}\text{O}_4\text{N}_2$  (apparently a nitrate: adduct of NO and  $\text{O}_2\text{N}_2$  to the double bond); catalytic reduction (as above) gave from 3 g. product: 2 g. oily material, identified as *hydroxyaminomaleic acid*,  $n_D^{20}$  1.4751,  $d_4^{20}$  0.9823, and 0.8 g. solid, sepd. into EtOH-sol. and -insol. fractions; the former was free of N, the latter, m.  $112-13^\circ$ , gave ppts. with silico- and phosphotungstic acids, gave a yellowish chloroplatinate, lost  $\text{NH}_3$  with KOH, and had the compn. of a mixt. of mono- and diaminomaleic acids. The mother liquor after the removal of the above-described primary reaction products was freed of solvent and gave 18.8 g. viscous oil,  $\text{C}_{11}\text{H}_{10}\text{NO}_4$ ,  $d_4^{20}$  1.0000,  $n_D^{20}$  1.0880, zero iodine no., had an acid reaction, and was quite unstable, losing N in the form of oxides even at room temp. Heating with excess 0.5 N aq. KOH indicated formation of  $\text{C}_{11}\text{H}_8\text{N}_2\text{O}_4\text{K}_2$ ; acidification gave a yellow oil,  $d_4^{20}$  1.0012,  $n_D^{20}$  1.0655, identified by analysis as the *nitrous ester of hydroxymaleic acid*,  $\text{C}_{11}\text{H}_{10}\text{O}_4\text{N}_2$ .

G. M. Kowalski

VASILYEV, S. V.

Sulphanilamide compounds of aromatic arsonic acids. II. Sulphanilamide derivatives of 3-amino-4-hydroxyphenylarsonic acid. S. V. Vasilyev (*J. gen. Chem., USSR*, 1949, 18, 311—316 (U.S. transl.)).—3-*p*-Acetamidobenzenesulphonamido- and 3-sulphanilamido-4-amino-phenylarsonic acid, and the arsenous-compound and arsenoxide corresponding to the last, have no, or only slight, activity against trypanosomes, relapsing fever, streptococcal infection, or experimental pneumonia. (See also A., 1950, 11, 339). D. P. Young.

cd

Sulfanilamide derivatives of aromatic arsonic acids  
 II. Sulfanilamide derivatives of 3-amino-4-hydroxybenzenearsonic acid. S. V. Vasilev. *Zhur. Obshch. Khim.* (J. Gen. Chem.) 19, 350-5 (1949); cf. C. I. 41, 951d. 3,4-H<sub>2</sub>N(HO)C<sub>6</sub>H<sub>3</sub>AsO<sub>3</sub>Na<sub>2</sub> (20 g.) in 200 ml. H<sub>2</sub>O treated over 25 min. with 20 g. p-AcNHCH<sub>2</sub>SO<sub>2</sub>Cl, kept 1.5 hrs. at 30-40°, filtered, and acidified gave 20.2 g. of 4,3-HO(p-AcNHCH<sub>2</sub>SO<sub>2</sub>NH)C<sub>6</sub>H<sub>3</sub>AsO<sub>3</sub>H<sub>2</sub>, m. 165-6°; hydrolysis of 20 g. by refluxing 1.3 hrs. in 25 ml. HCl (d. 1.067), cooling, and washing with H<sub>2</sub>O and cold EtOH gave the *deacetylated compd.* (I), m. 190-1° (decomp., from EtOH). Its Na salt (10 g.) in 300 ml. H<sub>2</sub>O treated gradually with 120 g. Na<sub>2</sub>SO<sub>3</sub> at 55° and cooled, gave 9.4 g. [4,3-HO(p-H<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>NH)C<sub>6</sub>H<sub>3</sub>AsO<sub>3</sub>Na] (II), which after Soxhlet extr. with EtOH, 20-fold diln. of the ext. with H<sub>2</sub>O, and addn. of HCl was obtained pure (0.7 g.), m. 180-1° (decomp.). I (10 g.) in 300 ml. concd. HCl, treated with SO<sub>2</sub> at 10-12° 2.75 hrs. and 50 ml. concd.

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HCl added, gave 4,3-HO(p-H<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>NH)C<sub>6</sub>H<sub>3</sub>AsO<sub>3</sub>H<sub>2</sub>·HCl, m. 102-3°; *free oxide*, by addn. of 10% Na<sub>2</sub>CO<sub>3</sub>, m. 135-6° (decomp., from EtOH). II (18 g. di-HCl salt) in 400 ml. H<sub>2</sub>O and 37.5 g. p-AcNHCH<sub>2</sub>SO<sub>2</sub>Cl, added gradually at 40°, gave after stirring 3.5 hrs. and filtration and washing with hot H<sub>2</sub>O, EtOH, Et<sub>2</sub>O, and (CH<sub>2</sub>Cl)<sub>2</sub>, 17.3 g. [4,3-HO(AcNHCH<sub>2</sub>SO<sub>2</sub>NH)C<sub>6</sub>H<sub>3</sub>AsO<sub>3</sub>H<sub>2</sub>] (III), 17.3 g. [4,3-HO(AcNHCH<sub>2</sub>SO<sub>2</sub>NH)C<sub>6</sub>H<sub>3</sub>AsO<sub>3</sub>H<sub>2</sub>] (III), decomp. 230-1°, which is hydrolyzed even at room temp. by dil. acids or bases. II (10 g.) in 100 ml. 1:5 HCl treated with calcd. amt. of Rongalite at 50-60° and kept 2.5 hrs., yields, after washing with dil. HCl, 4.2 g. 4,3-HO(p-H<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>NH)C<sub>6</sub>H<sub>3</sub>AsO<sub>3</sub>Na·4H<sub>2</sub>O (IV), decomp. 246-8°. The products were inactive or very weakly active in bactericidal tests. G. M. Kowaloff

CA

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**Mechanism of the electrolytic reduction of 3-amino-4-hydroxybenzenearsonic acid** S. V. Vasilev and G. D. Vovchenko. *Vestnik Mosk. Univ. Ser. Fiz.-Mat. i Estest. Nauk*, No. 2, 73-80 (1970). The reductions were carried out in a divided cell with porous porcelain cup separator, a C anode, and a Hg cathode. The best results are obtained with a catholyte of 12 g. arsonic acid in 100 ml. 3 N HCl, at 30-40°, vol. 3 amp.-sq. decm., and concd. HCl anolyte. Aliquots (2 ml.) were used as controls, and when titration required 7.9 ml. 0.1 N iodine soln. the reaction mixt. was worked up by filtration, acidification with 50 ml. concd. HCl, and filtration; the 3-amino-4-hydroxyphenylarsine oxide-HCl, so obtained decomp. 147° (from EtOH); free base (I) decompd. 131°. If the reduction is continued, the mixt. becomes viscous, then fluid once again; continuation until a 2-ml. aliquot requires 17.5 ml. 0.1 N iodine for titration, followed by addn. of 40 ml. concd. HCl, gave from 100 ml.

of soln. 2.2 g. 3,3'-diamino-4,4'-dihydroxyarsenobenzene (II). Continuation of the reduction until the titer was 20.7 ml., followed by neutralization with NaHCO<sub>3</sub>, gave 1.3 g. 3-amino-4-hydroxyphenylarsine (III), yielding a stable solid HCl salt (from aq. EtOH-HCl). Letting equal wts. of III and I stand 6 hrs. in 3 N HCl gave on acidification II, isolated as the di-HCl salt dihydrate. Similar reaction of III with the arsonic acid gave the same product; the filtrate with concd. HCl yields a colorless ppt., which with NaHCO<sub>3</sub> yields a solid, m. 130-4°; crystd. from EtOH, the 3-amino-4-hydroxyphenyldichloroarsine-HCl decomp. 146-8°. The electrolytic reduction of the arsonic acid thus goes through the following stages: RAsO<sub>3</sub>, (RAs-OH)<sub>2</sub>, (RAs)<sub>2</sub>O, RAs:AsR, (RAsH)<sub>2</sub>, and RAsH. G. M. Kosolapoff

1937



10

CA

**Action of nitrogen dioxide on ethyl cinnamate.** S. V. Vasil'ev and G. D. Vovchenko (M. L. Lomonosov Fine Chem. Tech., Moscow). *Zhur. Obshch. Khim.* (J. Gen. Chem.) 20, 1236-45 (1950).—The double bond of Et cinnamate (I) readily adds both gaseous and liquid ( $\text{NO}_2$ ). Soln. of 30 g. I in 75 ml. dry  $\text{Et}_2\text{O}$  at  $0^\circ$  with  $\text{N}_2\text{O}_4$  (64 g.) over 10 hrs., filtration after standing overnight, and evapn. gave 40.3 g. oil (II) and 4.5 g. cryst. product (III); the former was a bright yellow oil,  $d_{44}^{20} 1.3001$ ,  $n_D^{20} 1.4092$  (45° seems to be a misprint), which lost N oxides on storage and contact with moisture, while heating with dil.  $\text{H}_2\text{SO}_4$  yielded *nitrohydroxyhydrocinnamic acid*, a yellow oil, obtained only in crude state, and heating with alkali also severed one N atom. The results indicate that the II was a mixt. of 1,2-addn. products: *nitro-nitrite* and *dinitro ester*; in addn., reduction with Sn-HCl yielded a mixt. of *diamino-* and *hydroxyaminohydrocinnamic acids*, decomp.  $302-6^\circ$ , yielding the corresponding mixed HCl salts, m.  $143-9.5^\circ$ . III was blue in color, m.  $132-3^\circ$ , decomp.  $102^\circ$ , and its chem. behavior indicated the structure  $\text{Ph}(\text{CH}(\text{NO})\text{CH}(\text{ONO}_2)\text{CO}_2\text{Et})$ ; its reduction by Sn-HCl gave *aminohydroxyhydrocinnamic acid* (IV), does not m.  $328^\circ$ ; HCl salt, m.  $107-70^\circ$ . IV is probably the  $\beta$ -amino- $\alpha$ -hydroxy isomer, as the  $\alpha$ -amino- $\beta$ -hydroxy analog, prep'd. by reduction of the  $\text{N}_2\text{O}_4$  adduct to cinnamic acid, m.  $315-17^\circ$ , and forms an HCl salt, m.  $148-50^\circ$ . Addn. of 27 g.  $\text{N}_2\text{O}_4$  to 15 g. I in  $\text{Et}_2\text{O}$  over 5 hrs. at  $0^\circ$ , followed by standing overnight, as above, gave 7.0 g. oil, identical with II, and 7.6 g. solid, sepl. into 1.2 g. blue plates, m.  $132-3^\circ$  (identical with III), which on 7 months' storage lost color and changed to the corresponding *nitro-nitrite* (V), m.  $151-2^\circ$ ; the 2nd solid (6.4 g.), isolated by EtOH leaching, was a colorless solid, m.  $151-2^\circ$ , identified as V. its reduction by Sn-HCl gave IV, does not m.  $325^\circ$ .

G. M. Kowlapoff

CA

10

The action of nitrogen peroxide on the ethyl ester of cinnamic acid. S. V. Vasil'ev and G. D. Vovchenko. *J. Gen. Chem. U.S.S.R.* 20, 1283-92(1974)(Engl. translation). -- See *C.A.* 45, 1541e  
R. M. S.

VII-10 (1-5) J.V.  
 Action of nitrous oxide on diethyl maleate. S. V. Vasil'ev  
 and E. I. Nikitin (M. V. Lomonosov Fine Chem. Technol.  
 Inst., Moscow). *Sbornik Statei Obshchei Khim., Akad.  
 Nauk S.S.S.R.* 1, 305-10 (1953).—Passage of  $N_2O$  vapors  
 at 0° 8 hrs. into 36 g. di-Et maleate in  $Et_2O$ , kept overnight  
 at 0°, N oxides blown out with dry  $CO_2$ , and evapn. gave  
 49.8 g. brown oil,  $C_8H_{10}O_5N_2$ . The product on standing in a  
 desiccator decomp. with evolution of N oxides and formation  
 of a colorless solid (1.1 g.) which is free of N and which was  
 identified as tartaric acid. The residual yellow oil also de-  
 compd. on prolonged standing; hydrolysis of this oil either  
 with  $H_2O$  or with  $H_2SO_4$  gave a mixt. of products. Reduc-  
 tion of the oil with Sn-HCl gave 2.3 g. colorless  $C_8H_{10}O_5N_2$ ,  
 which did not melt at 320°, and which was identified pro-  
 visionally as a *aminohydroxysuccinic acid*; reduction of this  
 with H1-red P in sealed tube at 130-5° gave *aminosuccinic*  
*acid*, decomp. 256-7°, whose *Bz deriv.*, m. 179.5-80°. Be-  
 sides this product, the reduction yielded 1.8 g.  $C_8H_{10}O_5N_2$ ,  
 insol. in EtOH, m. above 310°, yields HCl salt, m. 239-  
 91.5° (decompn.); this material was identified as *diamino-*  
*succinic acid*. Therefore the initial product was a mixt. of  
*dinitrosuccinic acid di-Et ester*,  $EtO_2CCH(ONO)CH(NO_2)CO_2Et$ ,  
 and  $EtO_2CCH(ONO)CH(ONO)CO_2Et$ . When the re-  
 action was run in  $Et_2O$  with  $N_2O$  (with adequate cooling)  
 there were obtained similar results and the reaction mixt.  
 yielded: tartaric acid and a yellow unstable oil composed  
 primarily of the above pseudonitrosite and nitrite esters of  
 hydroxysuccinic acid derivs. Heating the product with  
 $H_2O$  yielded  $HNO_2$ . In all cases at 0° or 17° the maleate  
 was isomerized to fumarate in the course of the reaction;  
 thus the isomerization took place during addn. of the oxide  
 to the double bond.

—Chim. Org. Chem.

G. M. Kosolapoff

**"APPROVED FOR RELEASE: 08/31/2001**

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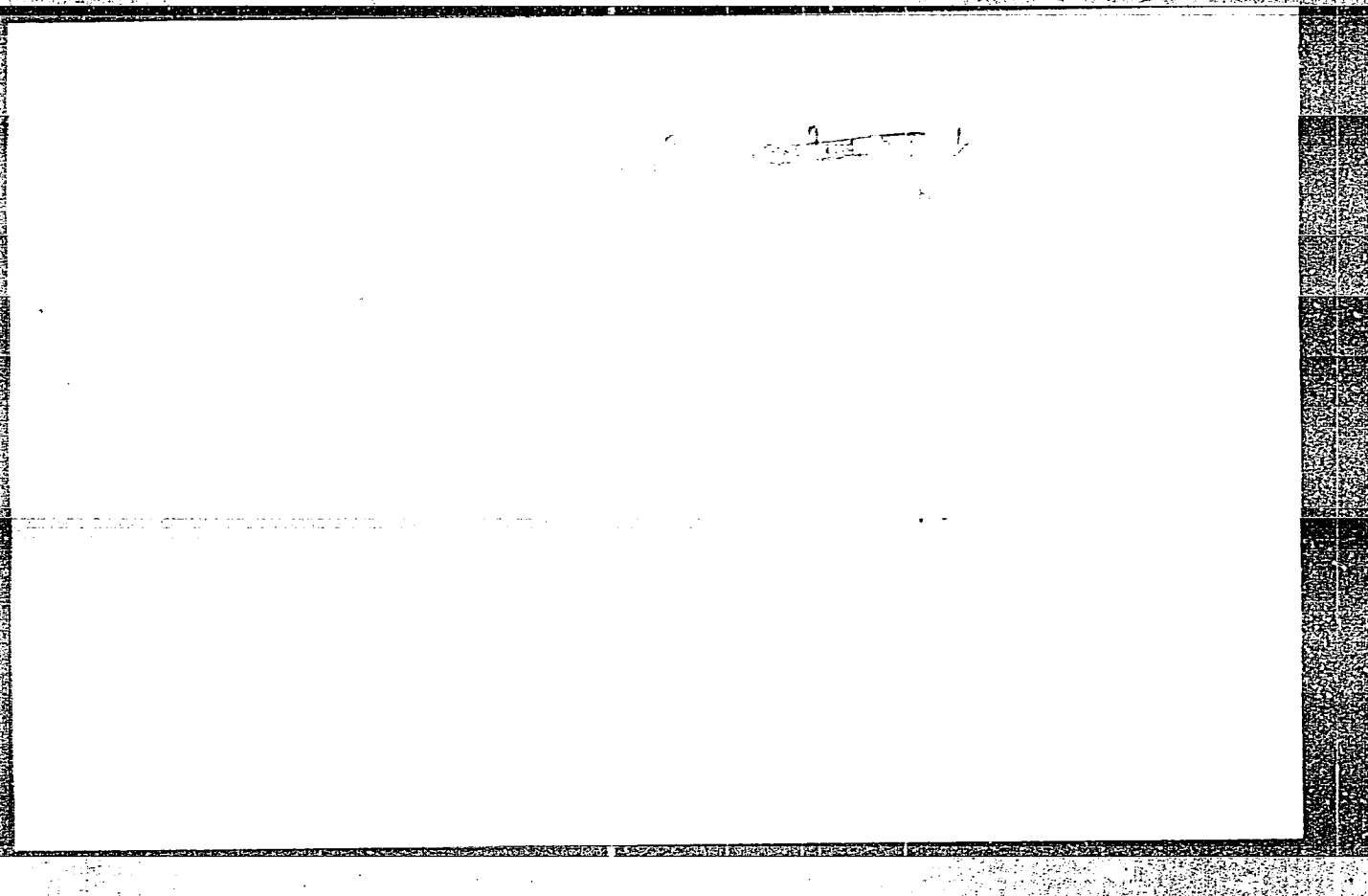
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APPROVED FOR RELEASE: 08/31/2001

CIA-RDP86-00513R001858920001-7"

VASIL'YEV, S. V., Doc Chem Sci (diss) -- "The effect of the oxides of nitrogen on unsaturated acids, ethers, and ketones". Moscow, 1959. 35 pp (Moscow Order of Lenin Agric Acad im K. A. Timiryazev), 115 copies (KL, No 22, 1959, 109)

VASIL'YEV, S.V.; MOCHALIN, V.B.; LIKHOSHERSTOV, V.M.

Ethers of substituted propargyl alcohols. Part 2: Effect of substituents in the alkylation reaction. Zhur. ob. khim. 34 no.10:3180-3183 O '64.

(MIRA 17:11)

1. Moskovskiy institut tonkoy khimicheskoy tekhnologii im. Lomonosova.



84882

S/079/60/030/010/024/030  
B001/B066

11.1170  
AUTHOR:

Vasil'yev, S. V.

TITLE:

Action of Nitrogen Tetroxide on Crotonic Acid 7

PERIODICAL:

Zhurnal obshchey khimii, 1960, Vol. 30, No. 10,  
pp. 3412 - 3414

TEXT: When ethylene and pseudobutylene are treated with nitrogen tetroxide, dinitro compounds result which, on reduction, are converted into diamines (Ref.1): The author studied the behavior of crotonic acid toward  $N_2O_4$ . If this reaction proceeds in the same way as with ethyl hydrocarbons, nitro compounds will be obtained, and on their reduction the corresponding amino acids. With an ether solution of crotonic or isocrotonic acid,  $N_2O_4$  gives the nitrogen ester of nitroxy-butyric acid. 7

Under the action of water on this compound, crystalline  $\alpha$ -nitro- $\beta$ -hydroxybutyric acid is formed. The amino acid and its hydrochloride were obtained by reducing nitroxy-butyric acid. Heating of the amino hydroxy-butyric acid with hydroiodic acid in a sealed tube gave  $\alpha$ -aminobutyric

Card 1/3

84882

Action of Nitrogen Tetroxide on Crotonic Acid S/079/60/030/010/024/030  
B001/B066

acid in the form of crystals. Furthermore, the author tried to determine the mutual conversions among crotonic and isocrotonic acids occurring under the action of  $N_2O_4$ . V. I. Yegorov assumed that the rearrangement occurs when the addition product is allowed to stand for some time (without verifying this assumption experimentally) (Ref. 2) (probably on the basis of the cis-trans rearrangement according to W. Wislicenus) (Ref.3). The present study of nitrogen tetroxide reactions, both with crotonic and isocrotonic acids, and also with other unsaturated acids, revealed that the behavior of the addition products on their formation and on standing does not confirm Wislicenus' opinion. The experiments carried out by the author indicated that, under the influence of both liquid and gaseous reagents, isocrotonic acid is converted to crotonic acid. Rearrangement in ether solution proceeded more rapidly than without a solvent. The double bond of crotonic acid is thus fully saturated by  $N_2O_4$ , and the groups  $ONO$  and  $NO_2$  are added (contrary to olefins which form dinitro compounds). There are 11 references: 9 Soviet and 2 German.

Card 2/3

84882

Action of Nitrogen Tetroxide on Crotonic  
Acid

S/079/60/030/010/024/030  
B001/B066

ASSOCIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii  
(Moscow Institute of Fine Chemical Technology)

SUBMITTED: October 17, 1959

Card 3/3

84883

S/079/60/030/010/025/030  
B001/B06611.1170  
AUTHORS:Vasil'ev, S. V., Zhuravleva, A. A., Kostomarova, V. I.,  
and Vasil'ev, G. S.

TITLE:

Effect of Nitrogen on Dibenzal Acetone 1

PERIODICAL:

Zhurnal obshchey khimii, 1960, Vol. 30, No. 10,  
pp. 3414 - 3416

TEXT: Proceeding from the reaction of nitrogen tetroxide with unsaturated aliphatic ketones, one of the authors (Ref.1) showed that, according to the structure of the initial ketone, addition products are obtained which differ as to nature and properties. The nitro group was found to be added to the least, and the ONO group to the most strongly hydrogenated carbon atom. When treating benzal acetone with nitrogen tetroxide, not only an addition to the double bond of the side chain takes place, but also a substitution of the hydrogen of the benzene ring in the para position. The behavior of dibenzal acetone toward nitrogen tetroxide was investigated. Dibenzal acetone dissolved in ether was treated with gaseous and liquid reagents. The nitrite of nitro

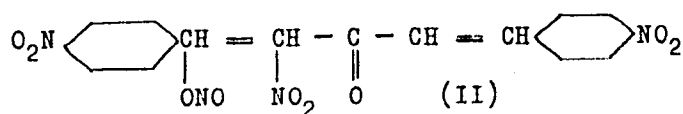
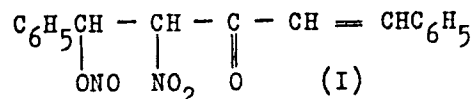
Card 1/3

84883

Effect of Nitrogen on Dibenzal Acetone

S/079/60/030/010/025/030  
B001/B066

oxyketone<sup>1</sup> (I) resulted in the former case, and the nitrite of trinitro  
oxyketone (II) in the latter.



By agitating with water, hydroxyl was substituted for the ONO group in both products (Refs. 2 and 3), to give the corresponding crystalline hydroxy-nitro-ketones. The addition products decomposed when heated with water or mineral acids on the water bath for 28-30 hours (Refs. 4 and 5). There are 5 references: 3 Soviet, 1 US, and 1 British.

Card 2/3

84883

Effect of Nitrogen on Dibenzal Acetone

S/079/60/030/010/025/030  
B001/B066

ASSOCIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii  
(Moscow Institute of Fine Chemical Technology)

SUBMITTED: March 16, 1959

Card 3/3

L 00309-66 EWT(1)/EPA(s)-2/EPF(n)-2/T-2/ETC(m) IJP(c) WJ

ACCESSION NR: AP5016659

UR/0382/65/000/002/0111/0122  
538.4+621.689

51  
B

AUTHOR: Vasil'yev, S. V.; Okhemenko, N. M.; Smirnova, L. G.

TITLE: Experimental investigation of the <sup>44.55</sup>magnetic fields <sup>44.55</sup>of an induction pump

SOURCE: Magnitnaya gidrodinamika, no. 2, 1965, 111-122 <sup>21, 94, 45</sup>

TOPIC TAGS: electromagnetic pump, magnetic field, magnetic induction

ABSTRACT: The spatial distribution of components of the magnetic induction,  $B$  in the gap between two-layer winding inductors of flat-type linear induction pump is studied. The test device operated at 80 KVA at current reading of 120 AMP. (operating without a load). The effects of loading on components of  $B$  are indicated. Both thermocouples on various metallic plates inserted in the gap and magnetic probe coils were used to obtain the data. Also given are data on measurement of integral distribution of components of  $B$ , demagnetization coefficient, and secondary leakage. Various fringe effects have been measured. In addition, it is shown that the above agrees with calculations which were carried out assuming the plane-parallel geometry for the magnetic field. Orig. art. has: 5 formulas, 9 figures, 1 table.

Card 1/2

L 00309-66

ACCESSION NR: AP5016659

ASSOCIATION: none

SUBMITTED: 30Jun64

ENCL: 00

SUB CODE: EM, ME

NO REF SOV: 008

OTHER: 000

Card

2/2



VASIL'YEV, S.V., *inh.*, *red.*; BRONSHTEYN, I.I., *red.*

[Experience in welding and sealing the joints and seams of precast reinforced concrete structural elements; collected studies] Opyt raboty po svarke i zadelke stykov i shvov sbornyykh zhelezobetonnykh konstrukttsii; sbornik materialov. Moscow, Energiya, 1964. 220 p. (MIRA 17:10)

1. Seminar inzhenerno-tekhnicheskikh rabotnikov i novatorov stroitel'stva teplovykh i gidroelektrostantsiy, Moscow, 1962.

VASIL'YEV, S.V.

Efficient design of ... Basin. Metallurg  
6 no.11:8 10 M ... (MIRA 14:11)

1. Novolipetskiy metallurgicheskiy zavod.  
(Blast furnaces, equipment and supplies)

KOROLEV, A.I.; BLINOV, S.T.; LUBNETS, I.A.; KOBURNEYEV, I.M.; TURUBINER, A.I.; VASIL'YEV, S.V.; CHERNENKO, M.A.; BELOV, I.V.; TELESOV, S.A.; MAZOV, V.F.; MEDVEDEV, V.A.; MAL'KOV, V.G.; BUL'SKIY, M.T.; TRUBETSKOV, K.M.; SHNEYEROV, Ya.A.; SLADKOSHTEYEV, V.T.; PALANT, V.I.; KUROCHKIN, B.N.; ZHDANOV, A.M.; BELIKOV, K.N.; SABIYEV, M.P.; GARBUZ, G.A.; PODGORETSKIY, A.A.; ALFEROV, K.S.; NOVOLODSKIY, P.I.; MOROZOV, A.N.; VASIL'YEV, A.N.; MARAKHOVSKIY, I.S.; MALAKH, A.V.; VERKHOVTSSEV, E.V.; AGAPOV, V.F.; VECHER, N.A.; PASTUKHOV, A.I.; BORODULIN, A.I.; VAYNSHTEYN, O.Ye.; ZHIGULIN, V.I.; DIKSHTSEYN, Ye.I.; KLIMASENKO, L.S.; KOTIN, A.S.; MOLOTKOV, N.A.; SIVERSKIY, M.V.; ZHIDETSKIY, D.P.; MIKHAYLETS, N.S.; SLEPKANEV, P.N.; ZAVODCHIKOV, N.G.; GUDENCHUK, V.A.; NAZAROV, P.M.; SAVOS'KIN, M.Ye.; NIKOLAYEV, A.S.

Reports (brief annotations). Biml. TSNIICM no.18/19:36-39 '57.  
(MIRA 11:4)

1. Magnitogorskiy metallurgicheskiy kombinat (for Korolev, Belikov, Agapov, Dikshteyn).
2. Kuznetskiy metallurgicheskiy kombinat (for Blinov, Vasil'yev, A.N., Borodulin, Klimasenko).
3. Chelyabinskiy metallurgicheskiy zavod (for Lubnets, Vaynshteyn).
4. Zavod im. Dzerzhinskogo (for Koburneyev).
5. Zavod "Zaporozhstal'" (for Turubiner, Mazov, Podgoretskiy, Marakhovski, Savos'kin).
6. Makeyevski metallurgicheskiy zavod (for Vasil'yev, S.V., Mal'kov, Zhidetskiy, Al'ferov).
7. Stal'proyekt (for Chernenko, Zhdanov, Zavodchikov).
8. VNIIT (for Belov).
9. Stalinskiy metallurgicheskiy zavod (for Telesov, Malakh).

(Continued on next card)

KOROLEV, A.I.---(continued) Card 2.

10. Nizhne-Tagil'skiy metallurgicheskii kombinat (for Medvedev, Novolodskiy, Vecher). 11. Zavod "Azovstal'" (for Bul'skiy, Slepkanov). 12. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy metallurgii (for Trubetskoy). 13. Ukrainskiy institut metallov (for Samoylov, Sladkovskiy, Kozlov). 14. Zavod "Krasnyy Oktiabr'" (for Palant). 15. Vsesoyuznyy nauchno-issledovatel'skiy institut metallurgicheskoy teplotekhniki (for Kurochkin). 16. Zavod im. Voroshilova (for Silyov). 17. Chelyabinskiy politekhnicheskii institut (for Morozov). 18. Giprostal' (for Garbuz). 19. Ural'skiy institut chernykh metallov (for Pastukhov). 20. Zavod im. Petrovskogo (for Zhigulin). 21. Ministerstvo chernoy metallurgii USSR (for Molotov, Siverakiy). 22. Glavspetsstal' Ministerstva chernoy metallurgii SSSR (for Nikolayev).  
(Open-hearth process)

137-58-6-11731

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 6, p 74 (USSR)

AUTHOR: Vasil'yev, S.V.

TITLE: Improvement of Open-hearth Furnace Design (Uluchsheniye konstruktsiy martenovskikh pechey)

PERIODICAL: Tr. Nauchno-tekhn. o-va chernoy metallurgii, 1957, Vol 18, pp 252-257

ABSTRACT: A description is offered of experience at the Makeyevka Plant. The life of the main roof was increased by increasing the number of hangers and improvement of the fastening to the beams thereof. After the bottom of the gas port was raised above the level of the sills, escape of slag into the gas slag pockets was eliminated, and this led to increasing the service life of the dividing wall to 2-2.5 years. The rear end beams were provided with water cooling. The power of the blowers was increased. In the substructure, the thickness of the regenerator walls and roofs was increased, the method of laying the checkers was changed (the Cowper system being installed), and a removable slag-pit design was tested. 1. Open hearth furnaces--Design  
Card 1/1 2. Open hearth furnaces--Materials M.M.

CA

*Electrochemistry*

**Electrolytic production of sulfuric acid.** S. V. Yasil'ev (Chuvash State Pedagog. Inst.). *Zhur. Priklad. Khim. (J. Applied Chem.)* 23, 385-95 (1950).—A considerable improvement of the process of Stender and Sirak (C.A. 20, 6149) for production of  $H_2SO_4$  by 3-compartment electrolysis of  $Na_2SO_4$  is obtained by the use of  $H_2WO_4$  as a heterogeneous catalyst for the anodic oxidation of the  $SO_2$  introduced into the anolyte. With  $H_2WO_4$ , the rate of absorption of  $SO_2$  in the acid  $Na_2SO_4$  soln. is approx. 2-3 times as great as in the presence of  $KMnO_4 + MnO_2$ . However, in an acid soln. of  $CuSO_4$ , absorption of  $SO_2$  is faster with  $KMnO_4 + MnO_2$  than with  $H_2WO_4$  as a catalyst. By cooling the  $H_2SO_4 + Na_2SO_4$  obtained by electrolysis to  $-10^\circ$  and applying combined electrolysis of solns. of  $Na_2SO_4$  and  $CuSO_4$

in the presence of catalysts and an uninterrupted stream of  $SO_2$  in the anolyte, one can obtain 20-30%  $H_2SO_4$  in the 1st stage and 47-53%  $H_2SO_4$  in the 2nd stage of the process. Current losses do not exceed 30-35%. N. Thou

VASIL'YEV, S. V. SAFCZHNIKOVA, R. I.

Chemistry - Study and Teaching

Reprocessing and utilizing silver wastes from photochemical laboratories for teaching purposes. Khim. v shkole no. 3, 1952.

9. Monthly List of Russian Accessions, Library of Congress, November 195<sup>6</sup><sub>2</sub>, Unclassified.

VASIL'YEV, S. V.

*[Handwritten mark]*

✓ Photochemical regeneration of silver from alkaline fixing solutions. S. V. Vasil'ev. *J. Appl. Chem. U.S.S.R.* 27, 247-51(1954)(Engl. translation).—See *C.A.* 48, 6244h.  
H. L. IL

*[Handwritten signature]*



4

Photorechemical regeneration of silver from alkaline fixing solutions

Chen, J. Chem. & Chem. Technol.

*[Handwritten signature]*

VASIL'YEV, Sergey Vasil'yevich; DMITRIYENKO, G.V., redaktor; PONOMAREVA, A.A., tekhnicheskiy redaktor

[Chemical experiments with electric current; manual for work outside class] Khimicheskie opyty s primeneniem elektricheskogo toka; rukovodstvo po vneklassnoi rabote. Moskva, Gos. uchebno-pedagog. izd-vo Ministerstva prosveshcheniia RSFSR, 1956. 110 p.  
(Electric currents) (MLBA 9:7)  
(Chemistry--Experiments)

VASIL'YEV, S.V. (g.Cheborsary Chuvashskoy ASSR)

Experiments on regeneration of silver from spent solution of  
fixer in a chemistry club. Khim. v shkole 12 no.2:61-64 Mr-Ap  
'57. (MLRA 10:3)  
(Silver) (Photographic chemistry)

VASIL'YEV, S. V.; CHEKUSHKIN, A. D.

Electrolyte for the preparation of metallic sodium. Khim. v  
shkole 17 no.6:65-66 N-D '62. (MIRA 16:1)

1. Pedagogicheskiy institut, g. Cheboksary.

(Electrolysis) (Chemistry--Experiments)  
(Sodium)

VASIL'YEV, S.V.; LAPSHINA, S.N.; KOSTOMAROVA, V.L.

Fatty oil from *Peucedanum ruthenicum*. Zhur.prikl.khim. 38  
no.9:2121-2123 S '65. (MIRA 18:11)

1. Moskovskiy institut tonkoy khimicheskoy tekhnologii imeni  
Lomonosova.

POKHVISNEV, A.N.; SHAROV, S.I.; ZHILKIN, N.K.; ORLOV, Yu.A.; MATVEYEV,  
P.M.; VASIL'YEV, S.V.; VIZLOV, Ye.M.

Operation of a 2,000 m<sup>3</sup> capacity blast furnace. Metallurg. 9  
no.1:7-11 Ja '64 (MIRA 18:1)

VASIL'YEV, S.V., inzh.; PETUKHOV, A.P., inzh.

Improving the granular composition of ski; coke. Steel and alloy:  
980-981 N '64. (MIP, 1964)

VESELOVSKAYA, T.K.; MACHINSKAYA, I.V.; BUTYUGIN, S.M., retsenzent;  
VASIL'YEV, S.V., retsenzent; BELOV, V.N., prof., red.  
[deceased]; FEDOROVA, T.P., red.; SHVETSOV, S.V., tekhn.  
red.

[Problems and exercises in organic chemistry] Zadachi up-  
razhneniia po organicheskoi khimii. Pod red. V.N.Belova.  
Petrozavodsk, Rosvuzizdat, 1963. 154 p. (MIRA 16:11)  
(Chemistry, Organic--Problems, exercises, etc.)



VASIL'YEV, S.Y.; ROZENBERG, B.I.; VINOGRADOV, V.A., red.; LARIONOV, G.Ye.,  
tekh.n.red.

[Electric equipment of peat enterprises] Elektrooborudovanie  
torfopredpriatii. Izd.2., perer. Moskva, Gos.energ.izd-vo,  
1960. 352 p. (MIRA 13:11)  
(Peat industry--Electric equipment)

VASIL'YEV, S.V.

Improving the distribution of materials in the blast furnace  
top. Metallurg 5 no.8:8-10 Ag '60. (MIRA 13:7)

1. Novolipetskiy metallurgicheskiy zavod.  
(Blast furnaces--Equipment and supplies)

KLIMENKO, G.A.; VASIL'YEV, S.Ye.

Computer laboratories attached to the Central Dispatch  
Administration of the power industry. Trudy Inst. elektrotekh.  
AN URSR no.19:125-135 '62. (MIRA 16:5)

(Electric power distribution)  
(Electronic computers)

KLIMENKO, G.A.; VASIL'YEV, S.Ye.

Work on the use of electronic digital computers in electric power systems carried out by a supporting group of Institute of Electrical Engineering at the Academy of Sciences of the U.S.S.R. attached to the Kiev Main Power Administration. Energ. i elektrotekh. prom. no.2:22-23 Ap-Je '62. (MIRA 15:6)

(Electric power distribution)  
(Electronic digital computers)

AVRAMENKO, V.N.; VASIL'YEV, S.Ye.; KLIMENKO, G.A.; KHRUSHCHOVA, Ye.V.

Use of digital computers for calculating load distribution  
efficiency between the electric power plants of the Kiev electric  
power system. Trudy Inst. elektrotekh. AN URSR no.19:5-15 '62.  
(MIRA 16:5)

(Electric power distribution)  
(Kiev Province— Electric power plants)

ZAKHAROV, N.N.; VASIL'YEV, T.I.; KVASHNIN, N.N.

The LBF paper filter for fine fuel cleaning. Avt.prom. no.1:  
43-44 Ja '60. (MIRA 13:5)

1. Gosudarstvennyy soyuznyy ordena Trudovogo Krasnogo Znameni  
nauchno-issledovatel'skogo avtomobil'nogo i avtomotornogo  
instituta.

(Filters and filtration)

VASIL'YEV, T. I.

USSR/Engineering - Oil filters

Card 1/1 : Pub. 12 - 4/16

Authors : Adamovich, A. V.; and Vasil'ev, T. I.

Title : Band paper filter for fine oil cleaning

Periodical : Avt. trakt. prom. 8, 12-15, Aug 1954

Abstract : The Scientific Automotive Institute designed and produced a band-paper filter of a new design, for fine oil cleaning in automobile and bus engines. Extensive tests were conducted on the above filter to determine its operation under various conditions, and a comparison is made with the existing oil filters type ASFO, LBF, and BMF. Tables; drawings; diagrams; illustrations; graphs.

Institution : .....

Submitted : .....

VASIL'YEV, T. V., kand. med. nauk

Annotated list of candidate and doctoral dissertations on  
syphilology presented during 1959. Vest. derm. i ven. no.6:  
77-82 '61. (MIRA 15:4)

1. Iz TSentral'nogo nauchno-issledovatel'skogo kozhno-venerolo-  
gicheskogo instituta (dir. - kandidat meditsinskikh nauk N. M.  
Turanov) Ministerstva zdravookhraneniya RSFSR.

(BIBLIOGRAPHY--SYPHILIS)



VASIL'YEV, T.V.

Summaries of candidate and doctoral dissertations on problems of  
dermatology and related sciences defended in 1960. Vest.derm. i  
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(BIBLIOGRAPHY--DERMATOLOGY)

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(BIBLIOGRAPHY---TRICHOMONIASIS)  
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(BIBLIOGRAPHY---URETHRA---DISEASES)

VASIL'YEV, T.V., kand. med. nauk

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affiliated sciences published in 1961. Vest. derm. i ven. 37  
no.8:75-85 Ag'63 (MIRA 17:4)

VASIL'YEV, T.V.

Secondary phenomena following lumbar puncture. Vest.ven.i derm. no.6:  
46-47 N-D '53. (MLRA 6:12)

1. Iz kliniki kozhnykh i venericheskikh bolezney (zaveduyushchiy -  
professor D.L.Voronov) Ryazanskogo meditsinskogo instituta im.  
I.P.Pavlova (direktor Ye.N.Kovalev).

(Spine--Puncture)